

Chemometric Analysis of Arbutin Derivatives from *Paederia foetida* and *Vitis vinifera* with Fourier Transform Infrared (FTIR)

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ABSTRACT

Introduction: *Paederia foetida* and *Vitis vinifera* were reported to have antioxidant activities due to their polyphenolic phytochemical contents. Arbutin may be included as active compounds in the plants. The chemometric analysis is used to identify the similarity of functional groups found in arbutin and that are in *Paederia foetida* and *Vitis vinifera*. **Materials and Method:** The ethanol extracts of *Paederia foetida* and *Vitis vinifera* were subjected to compound partition and characterization. Then a liquid-liquid partition is carried out using n-hexan and water solvents, followed by ethyl acetate and water partitions. This procedure is carried out continuously until the top layer is cleared. Partitioned compounds were analyzed using FTIR spectroscopy to identify functional groups. Furthermore, it was analyzed using the chemometric method (Principal Component Analysis and Cluster Analysis). **Results:** From the results of FTIR spectroscopy, functional groups in *Paederia foetida* and *Vitis vinifera* were identical to functional groups found in arbutin. PCA Analysis was achieved using principal component accounting an eigenvalue about 99,3 % of the total variance. **Conclusions:** CA analysis showed that compounds in *Paederia foetida* (Euclidean distance 0.537) and *Vitis Vinifera* (Euclidean distance 1.157) may be similar with arbutin.

Key words: Chemometrics, *Paederia foetida*, *Vitis vinifera*, Arbutin, PCA.

INTRODUCTION

Arbutin (β -arbutin) is a natural whitening compound of glycosylated derivatives from hydroquinone as tyrosinase inhibitors. This compound can be obtained from Ericaceae (berry, strawberry, huckleberry), Saxifragaceae, Asteraceae, Rosaceae, Lamiaceae, Apiaceae and is found in pears (*Pyrus communis* L.). The percentage of arbutin in plants varies greatly according to species (17% in *Arctostaphylos uva ursi* leaves, 5% in majorama leaves).¹

Several attempts have been made to investigate safe and effective tyrosinase inhibitors from both natural and synthetic compounds.²⁻⁴ Although there have been many studies and information about tyrosinase inhibitors, only few inhibitors can be applied due to their limitations in terms of cytotoxicity, selectivity and stability.^{5,6} Skunk vine plant (*Paederia foetida*) and Grape (*Vitis vinifera*) are included.

Paederia foetida are one of the plants mentioned as having arbutin and have never isolated arbutin from these plants. This plant contains chemical compounds in the stem and leaves, namely asperuloside, deacetylasperuloside, scandoside, paederosid, paederosidic acid, gamma-sitosterol, arbutin, oleanolic acid, and yawning oil. Grape plants (*Vitis vinifera*) have polyphenolic phytochemical content in the form of anthocyanins, tannins, flavonoids, resveratrol and phenolic acids. Polyphenols from grapes have a beneficial effect that can inhibit diseases such as heart disease, cancer, reduce plasma oxidation and slow aging.

In addition wine also has antioxidant, anticancer, anti-inflammatory, antiaging and antimicrobial effects. Another content contained in grape leaves is arbutin, mentioned that arbutin accumulates in grape leaves.^{7,8} It is necessary, therefore, to identify and characterize arbutin compounds contained in *Paederia foetida* and *Vitis vinifera* using FTIR spectrophotometers and chemometric analysis.

MATERIAL AND METHODS

Materials

Ethanol, ethyl acetate, chloroform, hexane, demineralized water and dimethyl sulfoxide (DMSO) were purchased from Merck, Indonesia. All chemicals and solvents were of analytical or pharmaceutical grade. Rotary Evaporatory (Buchi) for vaporization of the solvents. IR data were collected on a IRPrestige-21 using a KBr pellet.

Extraction and partition

Paederia foetida L. and *Vitis vinifera* were weighed and extracted separately by maceration method using ethanol. The dried samples were weighed and then put into the maceration container, then 96% ethanol was added to be completely submerged. The maceration container was closed and stored for 3 x 24 hours in a place that is protected from direct sunlight while stirring occasionally. Then extracts were filtered and separated from their pulps and filtrates. The ethanol filtrate obtained was then collected and evaporated using the rotary evaporator. The liquid-liquid partition was then carried out with n-hexane and water solvents. The n-hexan filtrate was collected and evaporated to obtain the dried extract. Then the

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water phase is placed back into the separating funnel and repartitioned with ethyl acetate and water. This procedure was repeated 2 to 3 times.

Identification of phenolic compounds

To identify arbutin compounds includes polyphenol test, the extracts were added with 1% FeCl₃ solution. Results are shown in the form of green, red, purple, dark blue, blue, blackish blue, or blackish green.⁹

Chemometric analysis

The Minitab software (Release 18; Minitab, State College, PA) was used for chemometric analysis including principal component analysis (PCA) and cluster analysis (CA) for the absorbance of FTIR spectra.

RESULTS AND DISCUSSION

FTIR Spectroscopy

FTIR spectroscopy was used to determine the functional groups of a compound. From the IR spectra the arbutin compounds were

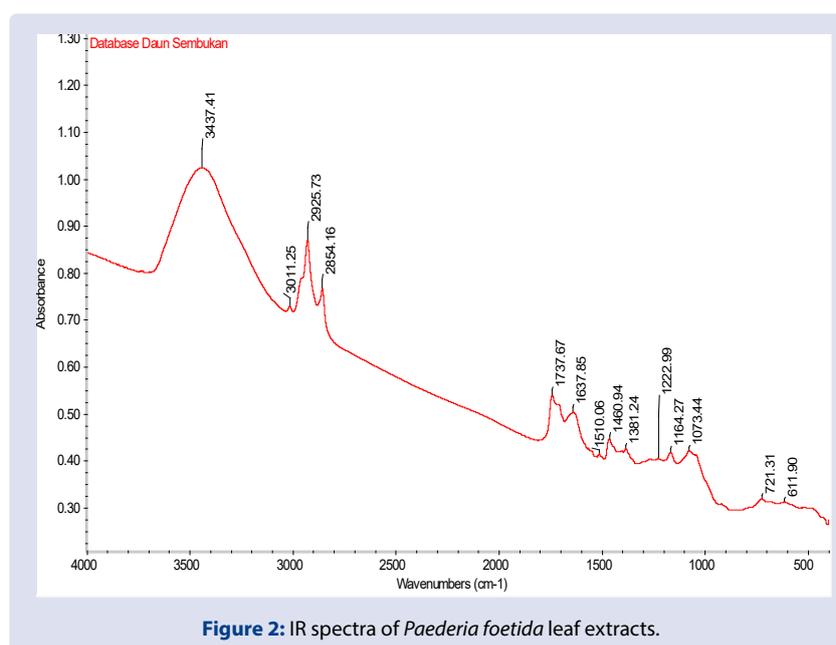
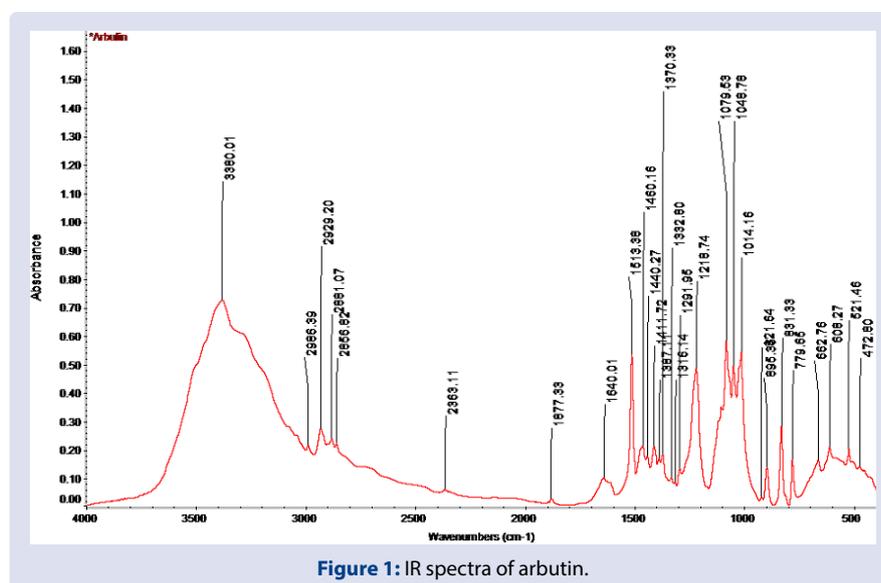
identified as hydroxyl groups, OH stretching (3380 cm⁻¹), CH stretching (2989, 2929 and 2881 cm⁻¹), and C-O stretching (1218 and 1291 cm⁻¹).

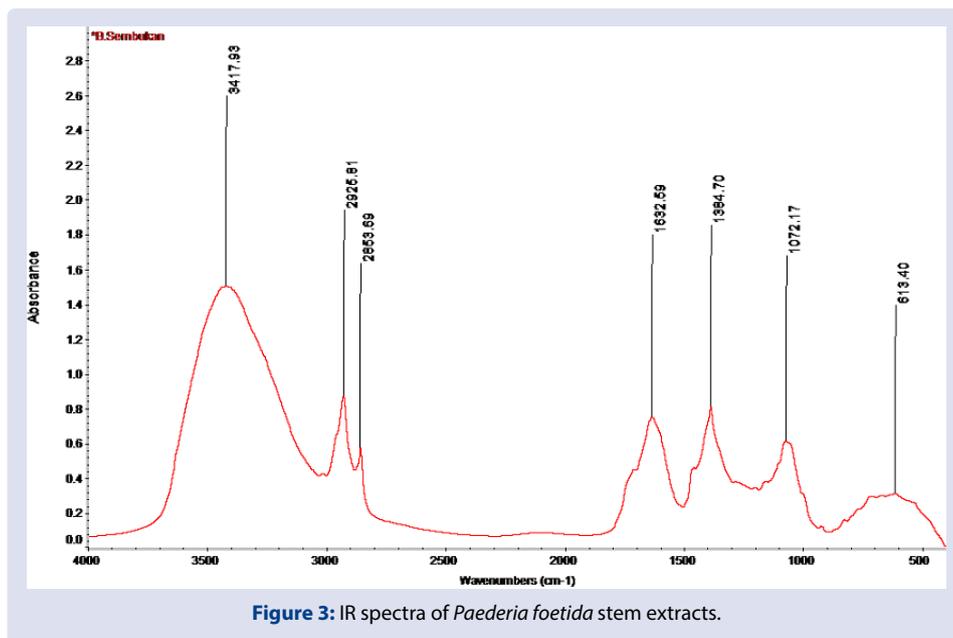
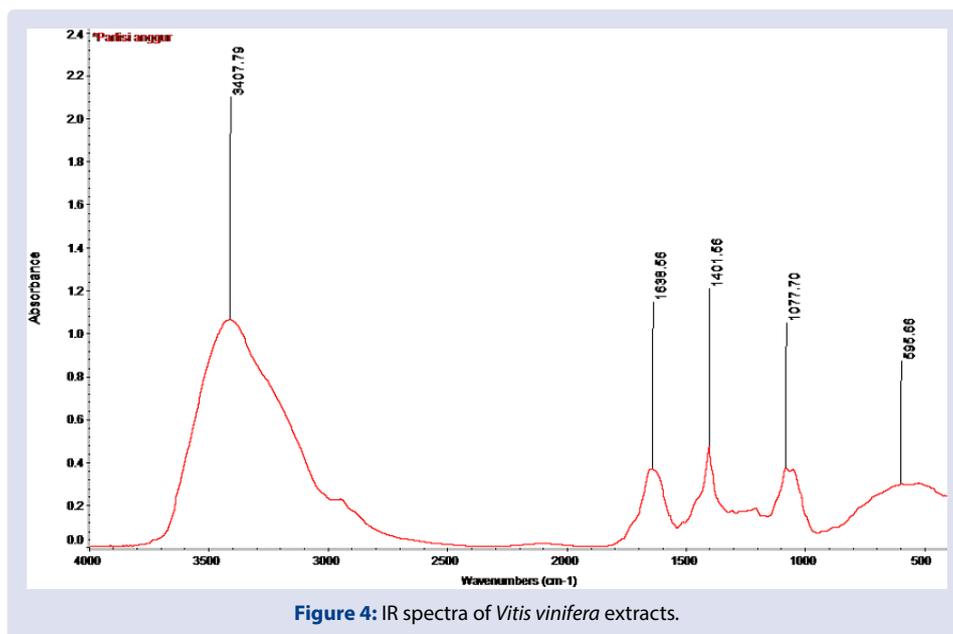
The IR spectra of *Paederia foetida* leaf extract were identified for hydroxyl groups, OH stretching (3437 and 3011 cm⁻¹), CH stretching (2925 and 2854 cm⁻¹), and C-O stretching (1222 and 1164 cm⁻¹). The IR spectra of *Paederia foetida* stem extract were identified for hydroxyl groups, OH stretching (3417 and 3011 cm⁻¹), CH stretching (2925 and 2853 cm⁻¹), and C-O stretching (1200 cm⁻¹).

For IR spectra of *Vitis vinifera* extract was identified for hydroxyl groups, OH stretching (3407 cm⁻¹), CH stretching (2990 cm⁻¹), and C-O stretching (1207 cm⁻¹).

Principal component analysis

Principal component analysis (PCA) is an unsupervised pattern recognition (Figures 1-4) method used in multivariate analysis.⁹ In this study, PCA was accomplished using FTIR spectra absorbances of 4 evaluated arbutin compounds at 4 frequencies as shown in Tables 1 and



Figure 3: IR spectra of *Paederia foetida* stem extracts.Figure 4: IR spectra of *Vitis vinifera* extracts.

2. PCA provided score and loading plot, showing the distribution of the samples and variables employed on the principal components (PCs), respectively. The score plot (Figure 5) exhibited patterns that may be correlated to sample characteristics. An eigenvalue of about 99.3% was achieved using two PCs. PC1 and PC2 described 95.3% and 4% of the variation, respectively. The study of the plot of loadings revealed distribution of variables and their correlations. From Figure 6, it is known that frequency regions at 1638.56 cm^{-1} and 3407.79 cm^{-1} make a larger contribution to the PCA model.

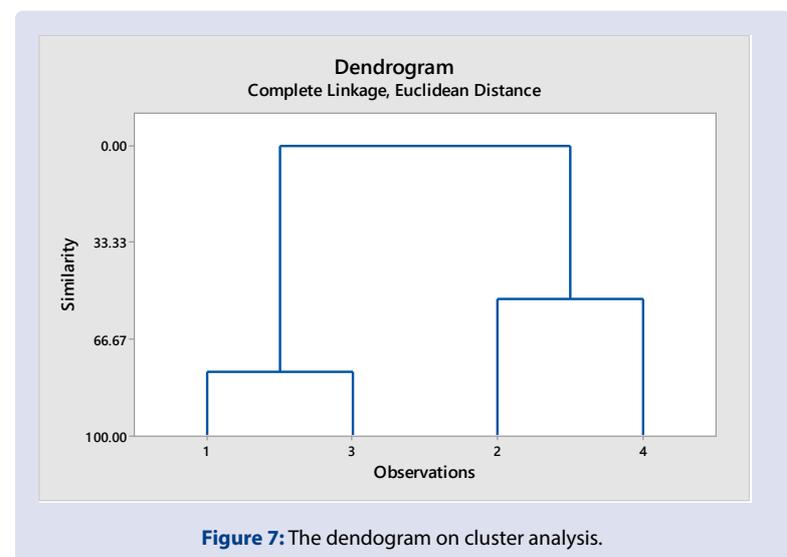
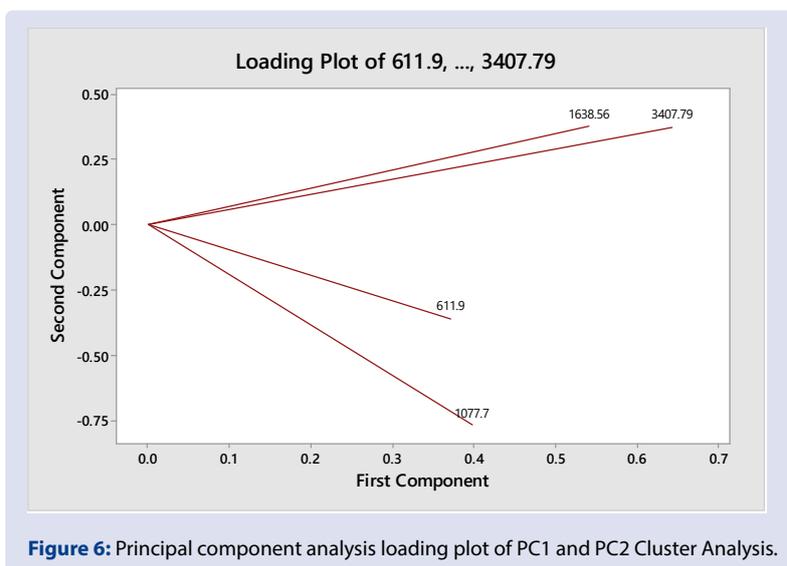
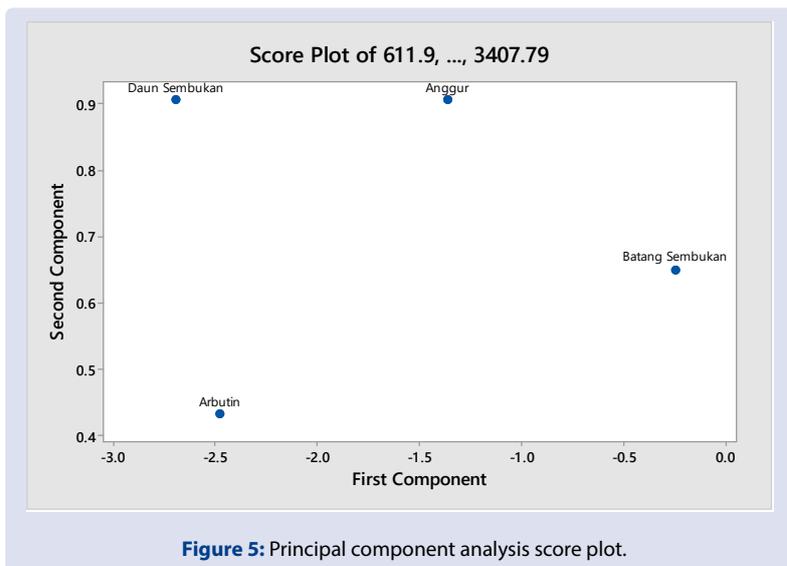
Cluster analysis (CA) is an algorithmic to construct hierarchy of clusters to assign observations to groups who are in the same class. In CA, clusters are visualized in dendrogram graphs. Figure 7 shows a dendrogram which is divided into two clusters: arbutin and *Paederia foetida* leaf with Euclidean distance 0.537, followed by *Paederia foetida* stem and *Vitis vinifera*. with Euclidean distance 1.157. In this study, CA results are in accordance with PCA results. They are in the same classes.

Table 1: IR spectra of compounds.

Arbutin	Wavenumbers (cm^{-1})			Functional Group
	<i>Paederia foetida</i> leaf	<i>Paederia foetida</i> stem	<i>Vitis vinifera</i>	
3380	3437, 3011	3417, 3011	3407	OH stretching
2986, 2929, 2881	2925, 2854	2925, 2853	2990	CH stretching
1218, 1291	1222, 1164	1200	1207	C-O stretching

Table 2: Peak intensities (absorbances) of compounds.

No	Compounds	Frequency			
		611.9	1077.7	1638.56	3407.79
1	Arbutin	-1.449	-1.131	-1.654	-0.931
2	<i>Vitis vinifera</i>	-1.214	-1.052	-0.937	0.018
3	<i>Paederia foetida</i> leaf	-1.69	-1.582	-1.498	-0.978
4	<i>Paederia foetida</i> stem	-0.693	-0.413	-0.29	0.508



CONCLUSIONS

In this work, we partitioned the extract of *Paederia foetida* and *Vitis vinifera* extracts. Furthermore, we combined FTIR spectroscopy with chemometric analysis to exhibit the similarity in their physico and chemical properties. CA analysis showed that arbutin derivatives may be included in *Paederia foetida* (Euclidean distance 0.537) and *Vitis Vinifera* (Euclidean distance 1.157).

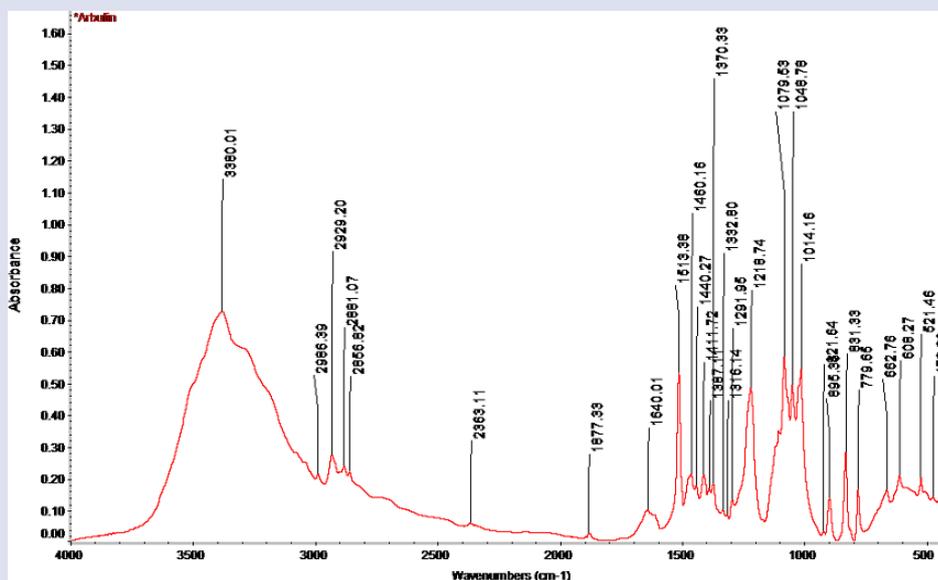
ACKNOWLEDGEMENTS

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GRAPHICAL ABSTRACT



SUMMARY

Arbutin compounds contained in *Paederia foetida* and *Vitis vinifera* were classified on the basis of their infrared spectra by pattern recognition method in chemometric analysis.

ABOUT AUTHORS



Yusnita Rifai was honored of Endeavour Post-Doctoral Fellowship 2014 to continue her post-doctoral study at the University of Newcastle Australia. Her research interests lie in the area of drug discovery, including drug synthesis. There are some of research in medicinal chemistry that obtained recognitions, one of them is from Timmerman Award in 2013.



Mukhriani is a postgraduate student at the Faculty of Pharmacy, Hasanuddin University. She is working on the research related to the chemical screening of naturally occurring arbutin from various Indonesian plants under the supervision of Yusnita Rifai.



Yulia Y Djabir is the head of clinical pharmacy laboratory at Faculty of Pharmacy Hasanuddin University. Her research interest is in the field of pharmacology and toxicology of synthetic and herbal medicine. She is also involved in some clinical studies regarding the safety use of drugs.



Gemini Alam is the head of phytochemistry laboratory at Faculty of Pharmacy Hasanuddin University. He is a professor in the field of natural products. His research is focused on the isolation, analysis, and bioactivity of natural products. Since 2017 he has been working on chemical screening using chemometric analysis.

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